

EE 403/503

Introduction to Plasma Processing



The kinetic Theory of Gases

(Elastc & Inelastc Collsions)

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Lecture Notes on Principles of Plasma Processing

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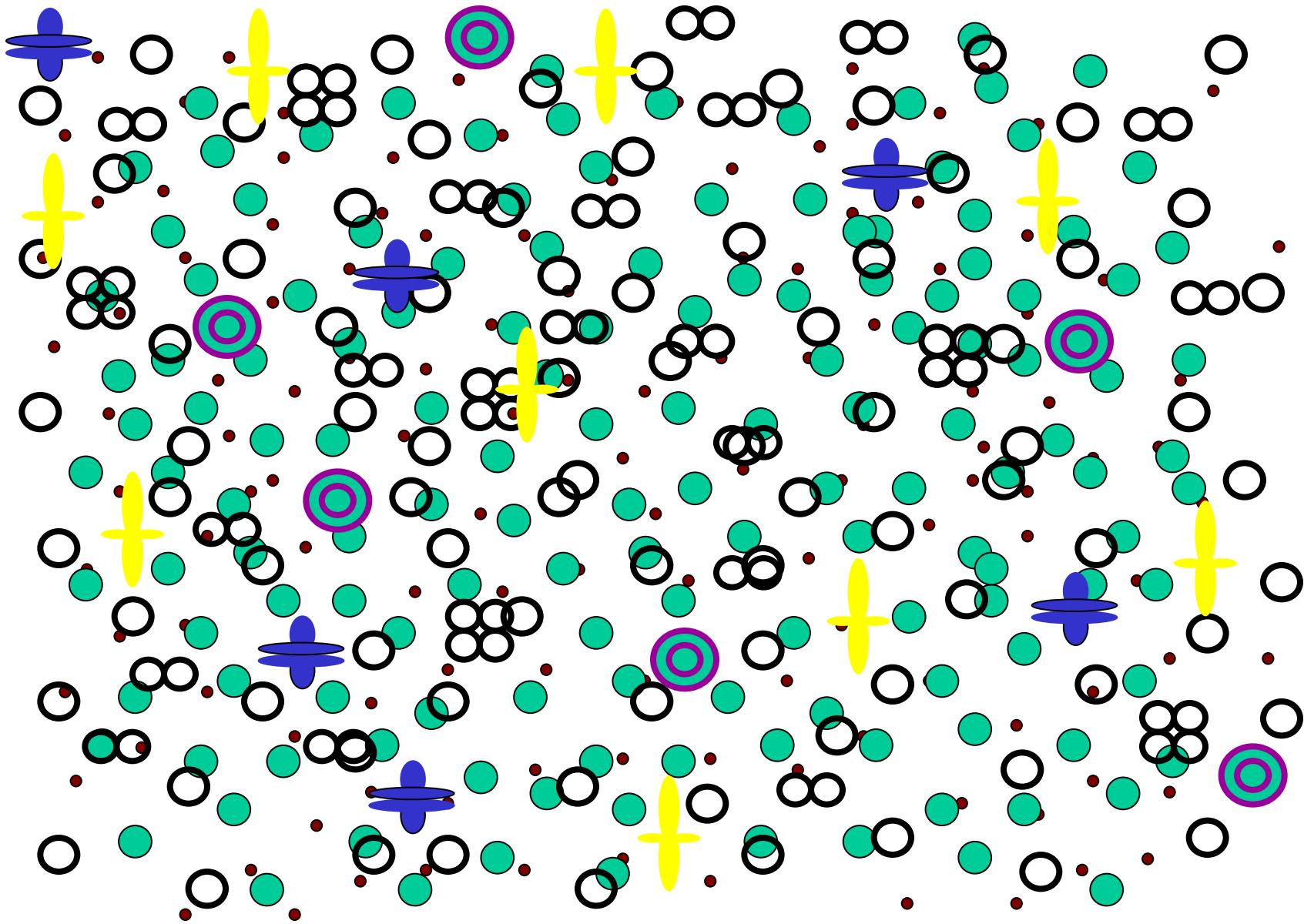
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Handbook of Advanced Plasma Processing Techniques

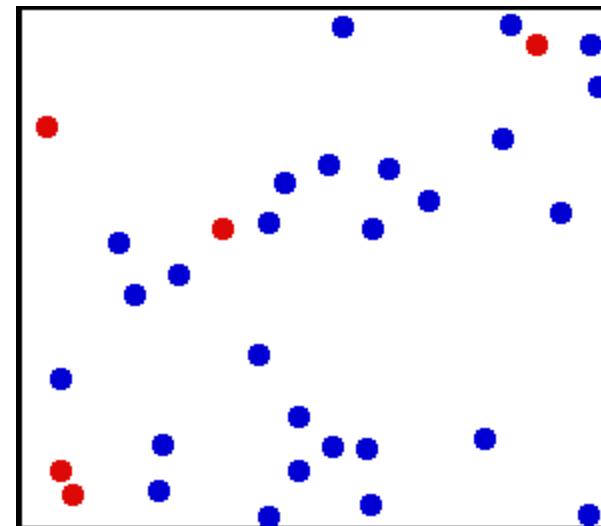
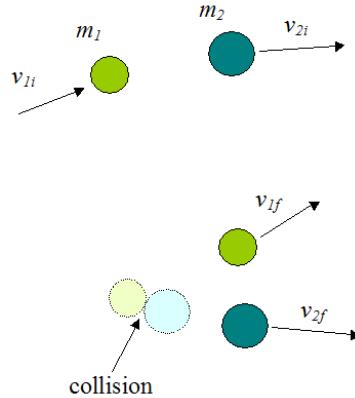
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Definition

Collision means the action of bodies striking or coming together (touching). When two particles pass each other and some measurable change in their path occurs, collision has taken place

Collisions involve forces (there is a change in velocity). Collisions can be *elastic*, meaning they conserve energy and momentum, *inelastic*, meaning they conserve momentum but not energy



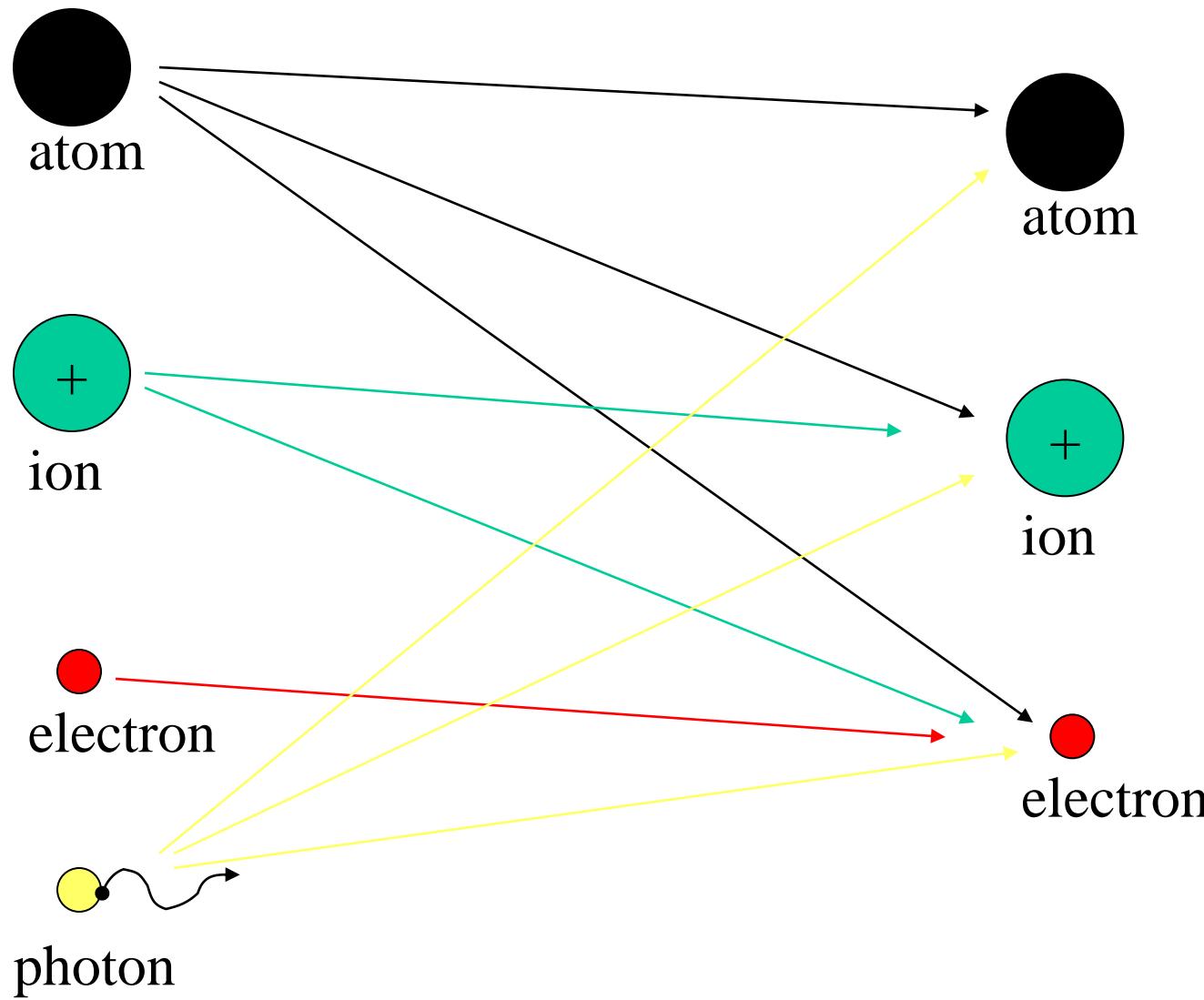
1- Type of Collisions

- a- Elastic Collisions
- b- Inelastic Collisions

2- Collision Parameters

3- A Simple Example

Binary Collisions



Major Types of Atomic Processes

(see the handout)

Particle Type	Symbol
Photon	o
Electron	e
Ground State atom or molecule	0
Electronically excited-one electron	0'
two electrons	0''
metastable state	0 ^m
Vibrationally excited molecules	0 ^v
Rotationally excited molecules	0 ^r
Positive ion ; singly, doubly, ...	1, 2, ...
Negative ion	$\bar{1}$

Examples:



10/01 Charge Transfer



o(00)/00'/00o

Photo Dissociation

Principal Kinds of Collision

Electrons



Ionization

Excitation

Penning Ionization

Elastic Scattering

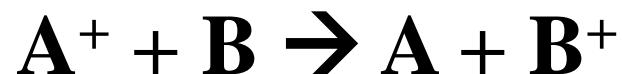
Dissociation

Dissociative Ionization

Dissociative Attachment

Recombination

Ions



Charge Exchange

Elastic Scattering

Ionization

Excitation

Recombination

Dissociation

Chemical Reaction

Photons



Photo Excitation
Photo Dissociation
Photo Ionization

Atoms



Elastic Scattering
Ionization
Dissociation
Photo Emission

	Energy [eV]
Ionization energy:	$e + N_2 = N_2^+ + e + e$ 15.6 $e + H_2 = H_2^+ + e + e$ 15.4 $e + SiH_4 = SiH_4^+ + e + e$ 11.4 $e + H = H^+ + e + e$ 13.6

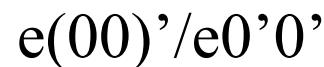
Electron Affinity:	$H^- = H + e$ 0.75 $OH^- = OH + e$ 18 $O_3^- = O_3 + e$ 2.0 $O_2^- = O_2 + e$ 0.44 $O^- = O + e$ 1.5 $SiH_3^- = SiH_3 + e$ 2.7
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Formation of multicharged negative ions is much less possible

See: “Plasma Physics and Engineering; Chapter 2 & 3; pp. 15-155

Particle Type	Sy mb ol
Photon	o
Electron	e
Ground State atom or molecule	0
Electronically excited-one electron	0'
two electrons	0''
metastable state	0 ^m
Vibrationally excited molecules	0 ^v
Rotationally excited molecules	0 ^r
Positive ion ; singly, doubly, ...	1, 2, ...
Negative ion	

Examples:



Electron impact dissociation



Bremsstrahlung, or free-free emission



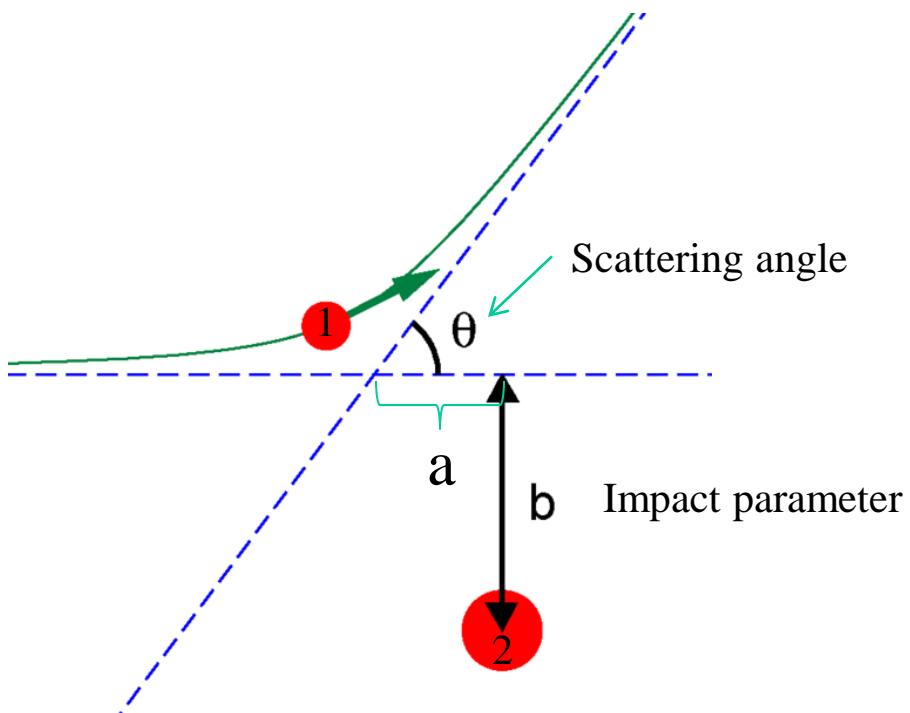
Vibrational excitation exchange



Electron impact dissociative ionization

Collision Parameters

1- Impact Parameter (b):

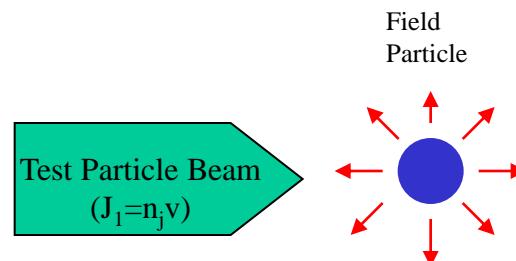


$$\theta = \pi - 2b \int_{r_{\min}}^{\infty} \frac{dr}{r^2 \sqrt{1 - (b/r)^2 - 2U/mv_{\infty}^2}}$$

$$\tan\left(\frac{\theta}{2}\right) = \frac{a}{b}$$

2- Total Collision Cross Section $Q_{12}(v)$:

$$Q_{12}(v) = \frac{\text{Number of test particles colliding with the field particles per unit time}}{\text{Flux Density } J_1}$$



$$Q_{12} = Q_{12}^{\text{elastic}} + Q_{12}^{\text{nonelastic}}$$

Elastic Collision: no changes in internal energy

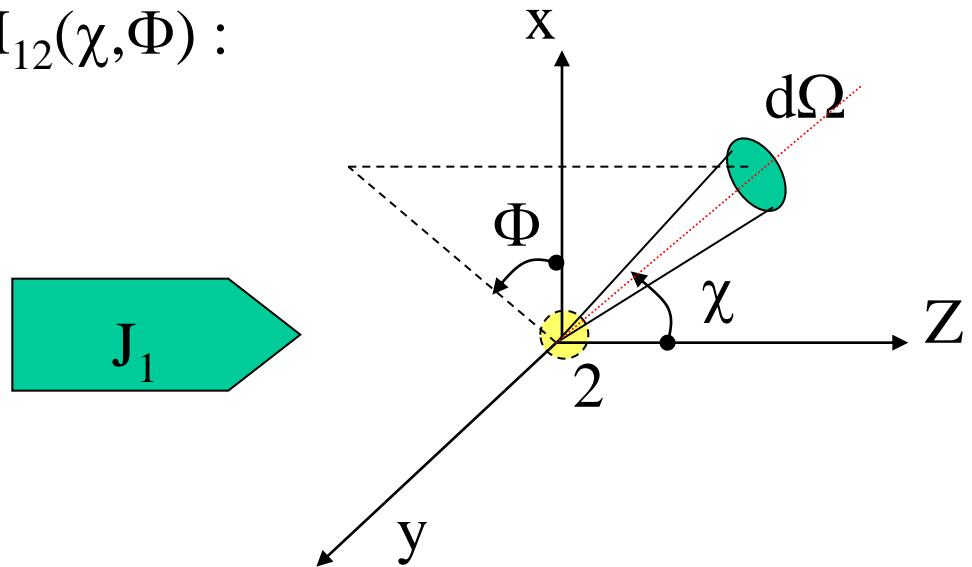
Inelastic Collision: with internal energy exchange

3- Nonelastic Collision Cross Section $Q_{12}(v)$:

$$Q_{12}^{(1 \rightarrow 2)}(v) = \frac{\text{Number of test particles causing a } (1 \rightarrow 2) \text{ excitation of the field particles per unit time}}{\text{Flux Density } J_1}$$

In an electron atom collision, kinetic energy of the electron needs to be above the energy of the first excited level in order to have an inelastic collision.

4- Differential Cross Sections $I_{12}(\chi, \Phi)$:



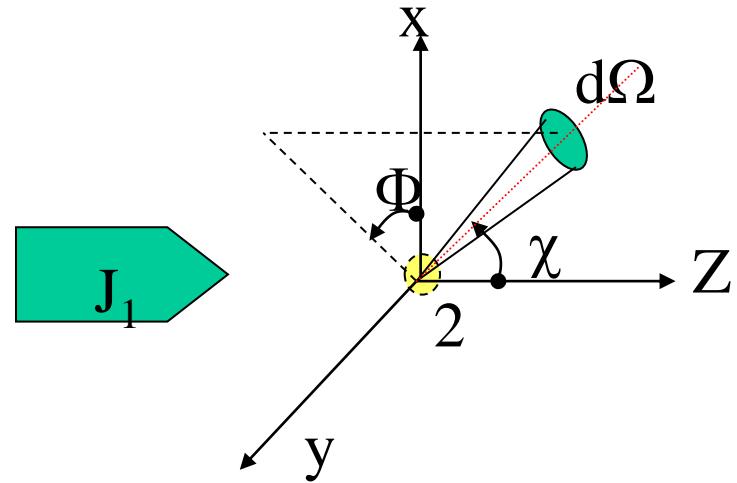
Number of test particles scattered by the field particles into the solid angle $d\Omega$ per unit time

$$I_{12}(\chi, \Phi) d\Omega = \frac{\text{Number of test particles scattered by the field particles into the solid angle } d\Omega \text{ per unit time}}{J_1}$$

$\nearrow Q_{12} = \int_{4\pi} I_{12}(\chi, \Phi) d\Omega$

Total Cross Section

5- Cross section for momentum transfer $Q_{12}^m(v)$:

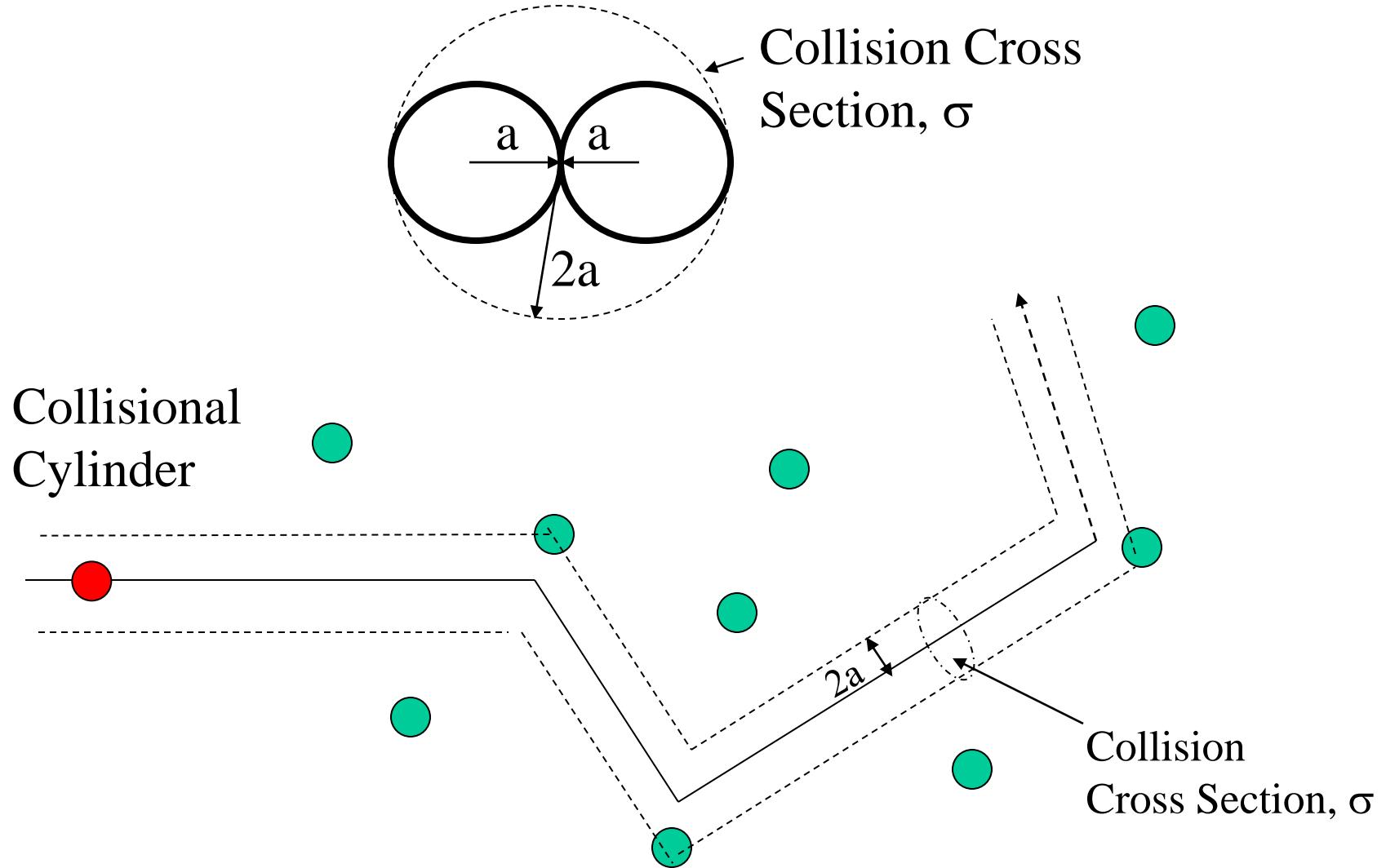


$$Q_{12}^m(v) \equiv \int_{4\pi} (1 - \cos \chi) I_{12}^{elastic}(\chi, \Phi) d\Omega$$

$(m_1 v J_1) Q_{12}^m(v)$ is the total loss in the z-component of momentum of the beam per unit time

$Q_{12}^m/Q_{12}^{elastic}$ the average fraction of momentum lost by the test particle in elastic collisions.

Binary Hard-sphere Collision



$$\text{Number of Collisions} = n \cdot \sigma \cdot v \cdot t$$

Reaction Rate Coefficient:

Reaction: A + B → C

$$\langle \sigma v \rangle = \frac{1}{n} \int_{-\infty}^{\infty} \sigma(v) v f(v) dv$$

Collision Cross Section
Velocity
Velocity distribution function

Collision Frequency

$$\nu_c = n \langle \sigma v \rangle$$

Density of the target

Interaction Rate

$$R = n^* v_c = n^* n \langle \sigma v \rangle$$

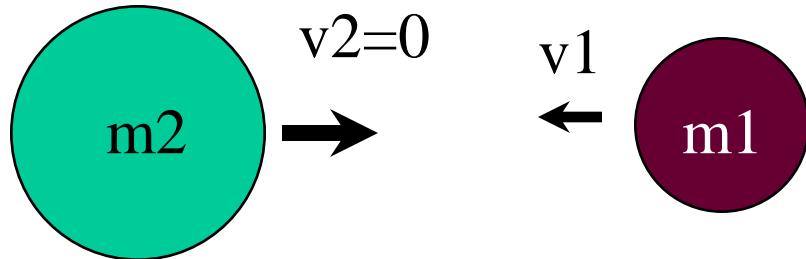
Density of the beam

Mean Free Path

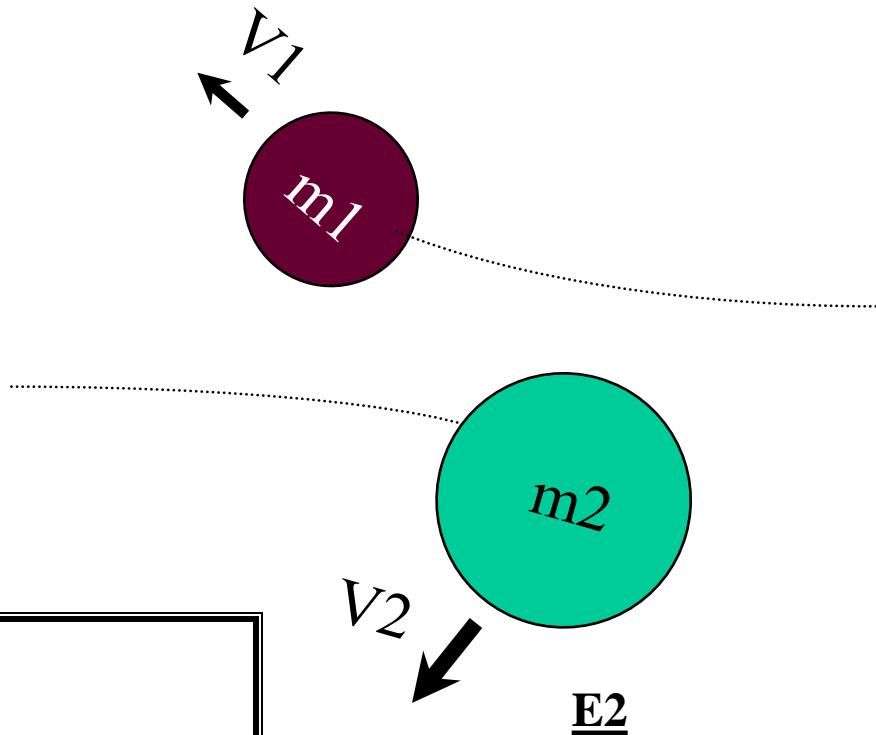
Average distance covered by a particle (photon, atom or molecule) between successive impacts. Distance at which the intensity of particles drops by $1/e$.

$$\lambda = \frac{\bar{v}}{n \langle \sigma v \rangle} \cong \frac{1}{\langle n \sigma \rangle}$$

Before the collision



After the collision



Conservation of Momentum:

$$m_1 v_1 = m_1 V_1 + m_2 V_2$$

Conservation of energy:

$$\frac{m_1 v_1^2}{2} = \frac{m_1 V_1^2}{2} + \frac{m_2 V_2^2}{2} + U$$



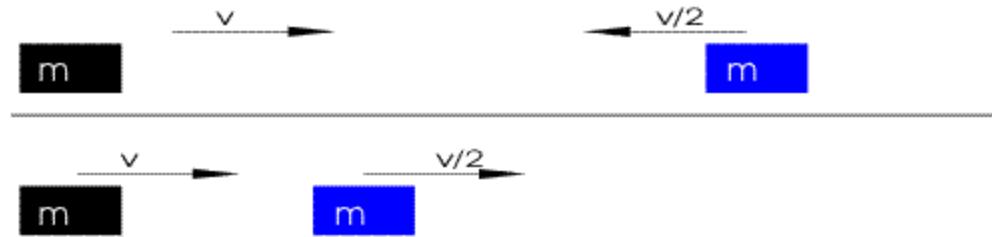
U=0 elastic collision

U≠0 inelastic collision

Elastic Collision (U=0)



Elastic collision of equal masses



Elastic collision of masses in a system with a moving frame of reference



Elastic collision of unequal masses

Elastic Collision (U=0)

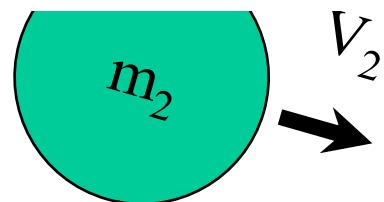
$$m_1 v_1 = m_1 V_1 + m_2 V_2$$

$$\frac{m_1 v_1^2}{2} = \frac{m_1 V_1^2}{2} + \frac{m_2 V_1^2}{2}$$

Fraction of energy transferred in an elastic collision

$$\frac{E_2}{E_1} = ?$$

r.



E2

Answer:

$$\frac{E_2}{E_1} = \frac{4 \times m_1 \times m_2}{(m_1 + m_2)^2}$$

E2

Inelastic Collision ($U \neq 0$)



E2

Elastic Collision (U=0)

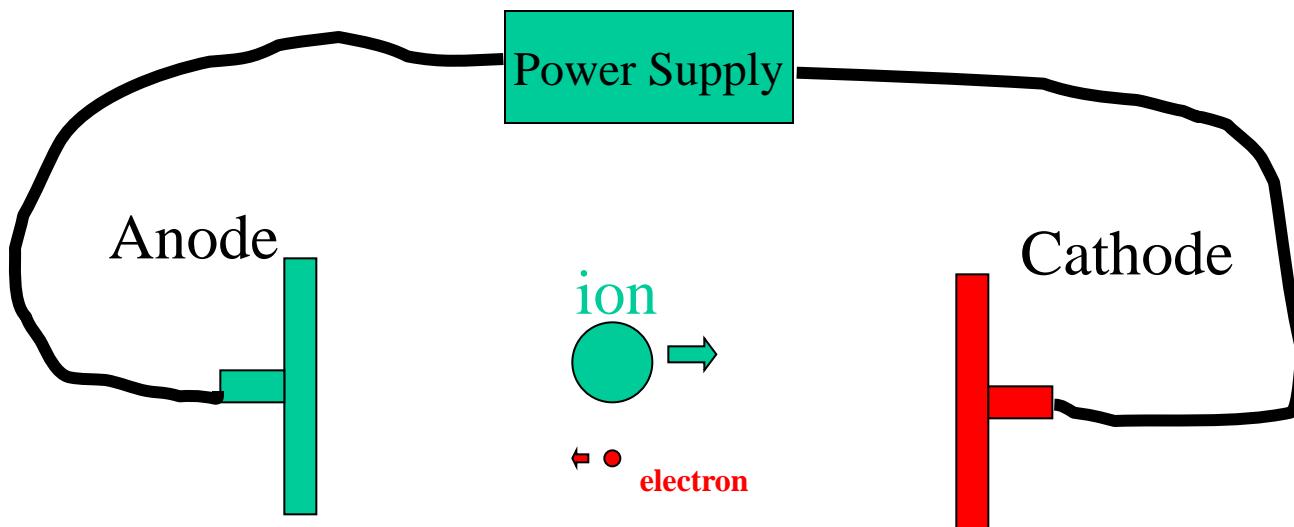
$$\frac{E_2}{E_1} = \frac{4 \times m_1 \times m_2}{(m_1 + m_2)^2}$$

Inelastic Collision (U=0)

$$U_{Max} = \frac{m_2}{m_1 + m_2} \times E_1$$

Case 1- $m_1=m_2$ $\left\{ \begin{array}{l} m_1=m_2=m_e \\ m_1=m_2=m_a \text{ or } m_i \end{array} \right. \rightarrow E_1 = E_2 \quad U_{max} \sim 0.5E_1$

Case 2- $m_2 \gg m_1$ $\left\{ \begin{array}{l} m_1=m_e \\ m_2=m_a \text{ or } m_i \end{array} \right. \rightarrow E_2 \sim 0 \quad U_{max} \sim E_1$



$$\text{Lorentz Force} \rightarrow F = eE \rightarrow F_i = F_e$$

$$\left\{ \begin{array}{l} F = ma \\ m_i \gg m_e \end{array} \right. \longrightarrow a_e \gg a_i \longrightarrow \text{Electrons accelerate faster and travel longer distances}$$

$F_i = F_e$ and electrons travel longer distances

Energy = Force x Distance

Electrons pick up more energy from the power source

Electrons pick up energy from the power supply



Electrons transfer energy to ions because of large collision cross sections (Coulomb interaction) compared to the atoms



Ions transfer energy to the neutral because of their approximately equal mass

Project

- Every week at the end of my lecture, we will discuss for ½ hour about the project.
- You will be randomly asked to present your most recent project activities and you will be graded.

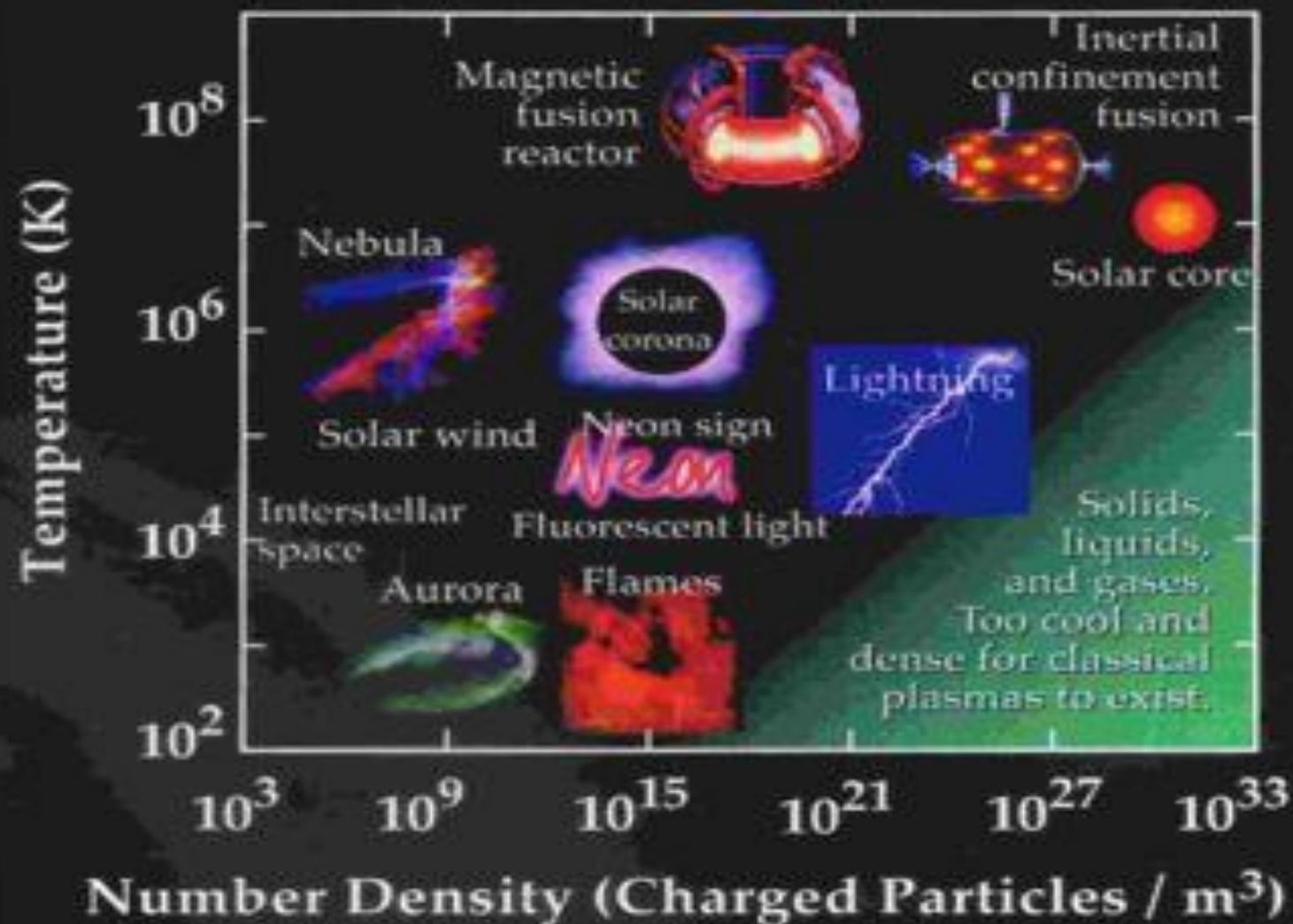
This week's task:

1-The objective (or importance) of your plasma application

2- At least two recent (max 1 year) activities (such as publication, Links to other research activities, ...) in your project area.

KIND OF PLASMA ?

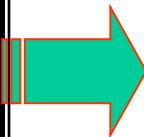
Plasmas - The 4th State of Matter



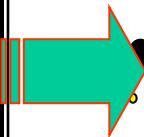
HOW DOES YOUR PLASMA DEVICE WORK?

Man-Made Plasmas

**Power
Source**



materials



**DC, AC, RF,
MW Power
Supplies**

Coupling

Capacitors

**Focused Laser
Beam**

**Electron and ion
beams**

Solid (e.g. iron, copper, ...)

Liquid (Mercury, ...)

Gas (Ar, H₂, SF₆, ...)

Breakdown

**Man-Made
Plasmas**

Maintenance