Problem Set 6, PS4 due June 2

- **PS6-1** 50 kg of water at 150 kpa and 25 C is contained in a piston cylinder device having a piston cross-sectional area of .1 m². The water is heated causing part of the water to vaporize. The piston reaches a linear spring having a spring constant of 100 kN/m when the volume contained by the piston cylinder reaches .2 m³. The piston then rises 20 cm further as more heat is added. Determine using the steam tables and EES a) the final pressure of the water, b) the work done during the processes. Sketch a schematic of the apparatus and a pressure volume property diagram of the processes.
- **PS6-2** A balloon material has the characteristic that the pressure inside the balloon is always proportional to the square of the diameter. If a spherical balloon made of this material contains 10 lbs of air at 30 psia and 800 R determine the work done when the volume of the balloon doubles as a result of heat transfer.

Problem Set 6, PS4 due June 2 p **PS6-1** $v_1 = v_f (a) 25^{\circ} C = .001003 \text{ m}^3/\text{kg}$ $V_1 = m \times v_1 = 50 \text{ kg} \times .001003 \text{ m}^3/\text{kg} = .05 \text{ m}^3$ a) $p_3 = p_2 + \frac{kx}{A} = 150 \text{ kPa} + \frac{100 \text{ N/m} \times .2 \text{ m}}{1 \text{ m}^2}$ $p_3 = 150 \text{ kPa} + 200 \text{ kPa} = 350 \text{ kPa}^{-1}$ 50 kg liquid water 150 kPa, 25 C $v_2 = \frac{.2 \text{ m}^3}{50 \text{ kg}} = .004 \text{ m}^3/\text{kg}$ V $V_3 = V_2 + .1 \text{ m}^2 \times .2 \text{ m} = .22 \text{ m}^3$ $v_3 = \frac{V_3}{m} = \frac{.22}{50} = .0044 \text{ m}^3/\text{kg}$ b)alternate = area under straight process line W = m p₁(v₂ - v₁) + $\frac{(p_3 + p_2)}{2} \times (V_3 - V_2)$ $T(a)P = 350 \text{ kPa and } v = .0044 \text{ m}^3/\text{kg}$ PressureTable = 138.88°C, in two phase region b) W = m $\int_{1}^{2} p \, dv + \int_{2}^{3} F \, dx = m p_1 (v_2 - v_1) + \int_{0}^{2m} (p_2 A + kx) \, dx$ W = 50×150(.004 - .001003) + $\frac{(350 + 150)}{2} \times (.22 - .20)$ W = m $p_1 (v_2 - v_1) + p_2 \times A \times x + \frac{k \times x^2}{2}$ W = 22.48 + 5 = 27.48 kJ W = 50×150(.004 - .001003)+150×.1×.2 + $\frac{100 \text{ N/m}}{2}$ (.2 m)² W = 22.48 + 3 + 2 = 27.48 kJ

"PS2 mae 204 Summer 2010"

"INPUT"

T1=25

p1=150

k=100

V2=.2

m=50

x23=.2

A=.1

"CALCULATION"

sv1=volume(Steam_IAPWS,T=T1,p=p1) V1=m*sv1 p3=p1+k*x23/A sv2=V2/m V3=V2+.1*x23 sv3=V2/m T3=temperature(Steam_IAPWS,v=v3,p=p3)

p2=p1 W1=m*p1*(sv2-sv1) w2=+p2*x23*A w3=(k*x23^2)/2

SOLUTION				
Unit Settings: [k	J]/[C]/[kPa]/[kg]/[degrees]		
A = 0.1	k = 100	m = 50	p1 = 150	p2 = 150
p3 = 350	sv1 = 0.001003	sv2 = 0.004	sv3 = 0.004	T1 = 25
T3 = 138.9	V1 = 0.05015	V2 = 0.2	V3 = 0.22	W1 = 22.48
w2 = 3	w3 = 2	x23 = 0.2		

Problem Set 6, PS4 due June 2 PS6-1

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$$\begin{array}{lll} p \propto D^2 & PS6-2 \\ p_1 = C \times D_1^2, & p = C \times D^2 \\ \hline p_1 = \frac{C \times D^2}{C \times D_1^2} & T_2 = \frac{p \ V_2}{m \ R} = \frac{144 \times 30 \times 1.587 = 47.6 \ psia}{T_2 = \frac{p \ V_2}{m \ R} = \frac{144 \times 30 \times 1.587 \times 197.6}{10 \times 53.35} = 2539.3 \ R \\ \Delta U = 10 \ lb_m \times 208 \ BTU/lb_m \ K \ (2539.3 - 800) = 36177 \\ V = \frac{\pi}{6} D^3 \ Volume \ of \ a \ sphere & .08 \ is \ the \ c_v \ at \ about \ average \ 1500^{\circ} \ F \\ Q = \Delta U + W = 36177. + 717 = 36,894. \ BTU \\ V_1 = \frac{mRT}{p} \\ P = p_1 \frac{V_1^2}{V_1^3} & V_1 = \frac{10 \ lb_m \times 53.35 \ ft \ lb_r / \ lb_m F \times 800 \ R}{144in^2/ft^2 \times 30 \ psia} \\ W = \int pdV = \int p_1 \frac{V_1^2}{V_1^3} V_1^2 & V_1^2 \\ W = \frac{10 \ lb_m \times 100}{V_1^3} V_1^2 V_1^3 & V_2 = 2 \times V_1 = 197.6 \ ft^3 \\ W = \frac{p_1}{V_1^3} \frac{3}{5} \left(\frac{V_2^3}{2} - V_1^{\frac{5}{3}} \right) \\ W = \frac{3 \times 30 \times 144}{98.8^{\frac{2}{3}} \times 5} \left(197.6^{\frac{5}{3}} - 98.8^{\frac{5}{3}} \right) / 778 \ \frac{ft \ lb_r}{BTU} & \frac{10 \ lb}{300 \ psia} \\ W = 717.8 \ BTU \end{array}$$