CE407 SEPARATIONS

Lecture 17

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No. of Ideal Stages (D Locate given points Lo, VN+1, LN @ Find M, LN -0 M+1 (3) Find VI Find (4 Find Points on up χ_{c} Vitl plot EQ, OP In (7) No. of stages LN 2



Multi-Stage Countercurrent Extraction Treybal pp. 452 Fig 1040 Minimum Entering Solvent Flow

- So far we have started with a given solvent flow, now we will see how to determine a reasonable flow
- Point #1
- Revisit the diagram for locating mixing point, M
- As the amount of solvent DECREASES
 - "M" moves toward L₀
 - V₁ moves to the left
 - Δ will move to the right
 - The line $\overline{V_1L_0}$ becomes steeper







- Point #2: Review Hunter-Nash method
- The # of steps are determined by alternating between:
 - Using Δ lines to do mass balances
 - Using tie lines to establish EQ relationships
- When the slopes of the Δ lines and tie lines are very different we make a lot of progress with each step
 - Similar to when OP lines and EQ curve are far apart
- When the slope of a Δ line is the same as the slope of a tie line we stop making progress
 - This is a pinch point
- The infinite number of steps corresponds to minimum solvent flow







- If we extend all of the relevant tie lines we see which leads to the furthest Δ location
 - Relevant tie lines are the those located between the tie line that passes through $\rm L_0$ and the one that passes through $\rm L_N$
- The Δ location furthest left corresponds to the largest flow that leads to a pinch point this is the Minimum Solvent Flow
 - Note that all smaller flows will have a pinch point, we are looking for one where you reach the point where there are no more pinch points
 - When Δ lies to left of triangle it is furthest out, when Δ lies to right of triangle it is closest
- If the tie lines all have similar slopes this will be the tie line that crosses at L_0 (Fig a)
- If the slopes vary, it could be a different tie line (Fig b)







- Label Leftmost intersection as Δ_{min}
- Notice the tie lines that are out of range are not used







- Now draw a line from Δ_{min} to L_0 and extend it to right hand side of phase boundary
- This determines V_{1,min}
- Note that this line is NOT necessarily a tie line
- Draw in $\overline{L_N V_{1,min}}$ and $\overline{L_0 V_{N+1}}$, their intersection determines **M**
- $\frac{(V_{N+1})_{min}}{L_0} = \frac{x_0 x_M}{x_M y_{N+1}}$ this gives the ratio of minimum solvent flow to feed flow





What Flow Should We Use?

- Same optimization as we did for other Unit Operations...
- Annual Cost = Depreciation + Solvent Cost



• Once again it turns out that it typically reaches a minimum at

$$(V_{N+1})_{opt} = 1.3(V_{N+1})_{min}$$





Batch Operation of a Stage



Think in terms of "before" and "after" the mixing and settling

- Charge Feed and Solvent to Vessel
- Mix thoroughly need proper hold time
- Stop agitation and let phase settle
- Aqueous phase is more dense and will be on bottom
- Drain material from bottom of vessel
 - First material is aqueous phase
 - Switch to another receiver when organic phase starts to come out



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Continuous Operation

Mixer-settlers



Think in terms of "flow in" and "flow out" each mixing and settling stage

- Continuous flow of Feed and Solvent to Mixing Vessel
- Mix thoroughly need proper residence time
- Mixture is continuously flowing to settler
- The two phases separate in settler and exit as two streams



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- Ellipse represents Stage 1
- L₀ is Feed into Stage 1
- V₂ is extract from Stage 2 feeding into Stage 1
- L₁ is raffinate flow leaving Stage 1
- V₁ is extract flow leaving Stage 1
- V₄ is solvent flow entering Stage 3, ie V_{N+1}
- L₃ is final raffinate flow exiting Stage 3, ie L_N



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Packed and plate towers



FIGURE 23.5

Perforated-plate extraction towers: (a) perforations in horizontal plates; (b) cascade weir tray with mixing and settling zones. (After Bushell and Fiocco.⁴) Very similar to the trays we have discussed in a Distillation Column

- Density difference is orders of magnitude lower than in a rectifying gas/liquid column (sp gr of 1 for aqueous and around 0.7 for organic)
- Both phases will be relatively high viscosity as opposed to the low viscosity vapor phase in distillation (velocities will be lower than in distillation column)
- Aqueous phase is more dense and will travel downward, organic phase will travel upward
- This means to location of the extract leaving the column depends on whether the extract is the aqueous phase or whether it is the organic phase

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<u>3</u>. A 500 kg/h feed stream with composition 45 mass % acetone (solute, C) and 55 mass % water (diluent, A) is to be contacted with trichloroethane (solvent, B) in a countercurrent liquid extraction battery. Entering trichloroethane is pure. The exiting raffinate should contain 20.2 mass % acetone on a trichloroethane-free basis. Our very good friend Elroy poses the following two questions:

(a) What is the minimum flow rate of trichloroethane required to achieve the desired composition of the exiting raffinate (corresponding to an infinite number of stages)?



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 $=\frac{\gamma_{0}-\gamma_{m}}{\gamma_{m}-\gamma_{m+1}}$ (VN+1)mir -0,45-0,38-0,194 $0_{138} - 0$ Lo= 500 $(V_{1/1})_{hi} = 96 / (8/h)$ $V_{1V+1} = 1.3 \times (V_{N+1})_{m'n}$ = 125 14



- <u>5</u>. A 450 kg/h feed stream with composition 38 mass % acetone (solute, C) and 62 mass % water (diluent, A) is to be contacted countercurrently with an MIK(solvent B)-rich solution of which the precise composition is 90 mass % MIK, 9 mass % acetone and the balance water. The exiting raffinate should contain 16.5 mass % acetone (C) and 83.5 mass % water (A) on an MIK(B)-free basis.
- (a) What is the minimum flow rate $(V_{N+1})_{\min}$ of the entering MIK-rich solvent stream required to achieve the desired separation (corresponding to an infinite number of stages)?







 $\frac{(V_{1\nu+1})_{min}}{L_{o}} = \frac{Y_{o} - X_{m}}{Y_{m} - Y_{n+1}}$ $= \frac{0.38 - 0.31}{0.31 - 0.09}$ = 0.33 $(V_{N+1})_{m_{1}} = 455 \times 0.33 \times 150$ $V_{1+1} = 1.3 \times = 195$

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